





## **CBSE Class 10 Science Sample Paper 2023-24**

Sample Question Paper 2023-24 Class X Science (Subject Code – 086)

Max. Marks: 80 Time Allowed: 3 hours

### **General Instructions:**

- i. This question paper consists of 39 questions in 5 sections.
- ii. All questions are compulsory. However, an internal choice is provided in some questions. A student is expected to attempt only one of these questions.
- iii. Section A consists of 20 objective type questions carrying 1 mark each.
- iv. Section B consists of 6 Very Short questions carrying 02 marks each. Answers to these questions should be in the range of 30 to 50 words.
- v. Section C consists of 7 Short Answer type questions carrying 03 marks each. Answers to these questions should be in the range of 50 to 80 words.
- vi. Section D consists of 3 Long Answer type questions carrying 05 marks each. Answer to these questions should be in the range of 80 to 120 words.
- vii. Section E consists of 3 source-based/case-based units of assessment of 04 marks each with sub-parts.

#### **Section-A**

Select and write the most appropriate option out of the four options given for each of the questions 1 - 20. There is no negative mark for incorrect response.

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Q. Nos.	Questions	Marks
1	Test tube containing solution of sodium sulphate  Test tube containing solution of barium chloride	
	Identify the product which represents the solid state in the above reaction.	
	a) Barium chloride	
	b) Barium sulphate	
	c) Sodium chloride	
	d) Sodium sulphate	
2	The colour of the solution observed after 30 minutes of placing zinc metal to copper sulphate solution is	1
	a) Blue	
	b) Colourless	
	c) Dirty green	
	d) Reddish Brown	

		1				
3	Mild non-corrosive basic salt is	1				
	a) $Ca (OH)_2$					
	b) NaCl					
	c) NaOH					
	d) NaHCO <sub>3</sub>					
4	On adding dilute sulphuric acid to a test tube containing a metal 'X', a colourless gas is produced when a burning match stick is brought near it. Which of the following correctly represents metal 'X'?  a) Sodium  b) Zinc  c) Copper  d) Silver	1				
5	Which one of the following correctly represents Sodium oxide?	1				
	a) $Na^{+2} = \begin{bmatrix} x \times x \\ x \times x \\ x \times x \end{bmatrix}^{-2}$ b) $2Na^{+} \begin{bmatrix} x \times x \\ x \times x \\ x \times x \end{bmatrix}^{-2}$					
	c) $2 \text{ Na}^{\frac{1}{2}} 2 \begin{bmatrix} x \times x \\ x & 0 \times x \\ x \times x & x \end{bmatrix}^{-1}$ c) $\text{Na}^{\frac{1}{2}} \begin{bmatrix} x \times x \\ x & 0 \times x \\ x \times x & x \end{bmatrix}^{-2}$					
6	An element with atomic number will form a basic oxide.	1				
	a) 7 (2,5)					
	b) 17 (2,8,7)					
	c) 14 (2,8,4)					
	d) 11 (2,8,1)					
7	An element 'M' has 50% of the electrons filled in the 3 <sup>rd</sup> shell as in the 2nd shell. The atomic number of 'M' is:	1				
	a) 10					
	b) 12					
	c) 14					
	d) 18					
8	Generally food is broken and absorbed within the body of organisms. In which of the	1				
	following organisms is it done outside the body?					
	<ul><li>a) Amoeba</li><li>b) Mushroom</li></ul>					
	c) Paramoecium					
	d) Lice					
9		1				
	Receptors are usually located in sense organs. Gustatory receptors are present in	1				
	<ul><li>a) tongue</li><li>b) nose</li></ul>					
	b) nose c) eye					
	d) ear					
	2					
	2					







a) DNA which is directly influenced by growth hormone. b) Genes which regulate the proteins directly. c) Growth hormones under the influence of the enzymes coded by a gene. d) Growth hormones directly under the influence a gene.  12  A sportsman, after a long break of his routine exercise, suffered muscular cramps during a heavy exercise session. This happened due to: a) lack of carbon dioxide and formation of pyruvate. b) presence of oxygen and formation of ethanol. c) lack of oxygen and formation of lactic acid. d) lack of oxygen and formation of carbon dioxide.  13  An object is placed in front of a convex mirror. Its image is formed: a) at a distance equal to the object distance in front of the mirror. b) at twice the distance of the object in front of the mirror. c) half the distance of the object in front of the mirror. d) behind the mirror and it's position varies according to the object distance.  14  When light enters the atmosphere it strikes on extremely fine particles, which deflect the rays of light in all possible directions, This is due to - a) reflection of light b) atmospheric refraction c) scattering of light d) dispersion of light	
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In 1987, an agreement was formulated by the United Nations Environment Programme (UNEP) to freeze the production of "X" to prevent depletion of "Y". "X" and "Y" respectively referred here are:	1
a) Ozone; CFCs	
b) CFCs; rays UV	
c) CFCs; Ozone	
d) UV rays; Diatomic oxygen	
Which of the following features relates to biodegradable substances?	l
a) Broken down by biological processes	
b) Remain inert	
c) Persist in environment for long time	
-,	







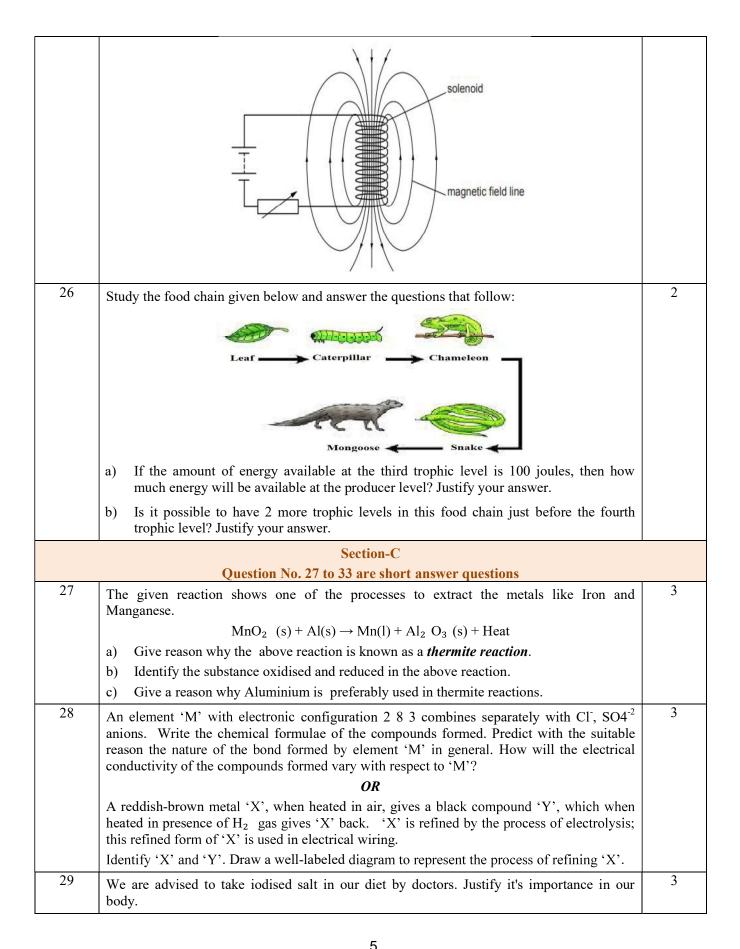


	Question No. 17 to 20 consist of two statements – Assertion (A) and Reason (R). Answer these questions selecting the appropriate option given below:		
	a) Both A and R are true, and R is the correct explanation of A.		
	b) Both A and R are true, and R is not the correct explanation of A.		
	c) A is true but R is false.		
	d) A is false but R is true.		
17	Assertion: Rusting of Iron is endothermic in nature.	1	
	Reason: As the reaction is slow, the release of heat is barely evident.		
18			
10	Assertion: Probability of survival of an organism produced through sexual reproduction is more than that of organism produced through asexual mode.		
	Reason: Variations provide advantages to individuals for survival.		
19	Assertion: A compass needle is placed near a current carrying wire. The deflection of the compass needle decreases when the magnitude of the current in the wire is increased.	1	
	Reason: The strength of a magnetic field at a point near the conductor increases on increasing the current.		
20	Assertion: Biodegradable substances result in the formation of compost and natural replenishment.	1	
	Reason: It is due to breakdown of complex inorganic substances into simple organic substances.		
	Section-B		
	Question No. 21 to 26 are very short answer questions		
21	Dil. HCl is added to Zn granules." How will you prove that chemical change has taken place here? Support your response with two arguments.	2	
22	State the post-fertilisation changes that lead to fruit formation in plants.	2	
23	What is the purpose of making urine in the human body? Name the organs that stores and releases the urine.	2	
	OR		
	Why do arteries have thick and elastic walls whereas veins have valves?		
24	The refractive indices of three media are given below:    Medium	2	
	A ray of light is travelling from A to B and another ray is travelling from B to C.		
	(a) In which of the two cases the refracted ray bends towards the normal?		
	(b) In which case does the speed of light increase in the second medium?		
	Give reasons for your answer.		
25	A piece of wire of resistance R is cut into three equal parts. These parts are then connected in parallel. If the equivalent resistance of this parallel combination is $R_1$ , what is the value of the ratio $R_1$ : R?	2	
	OR		















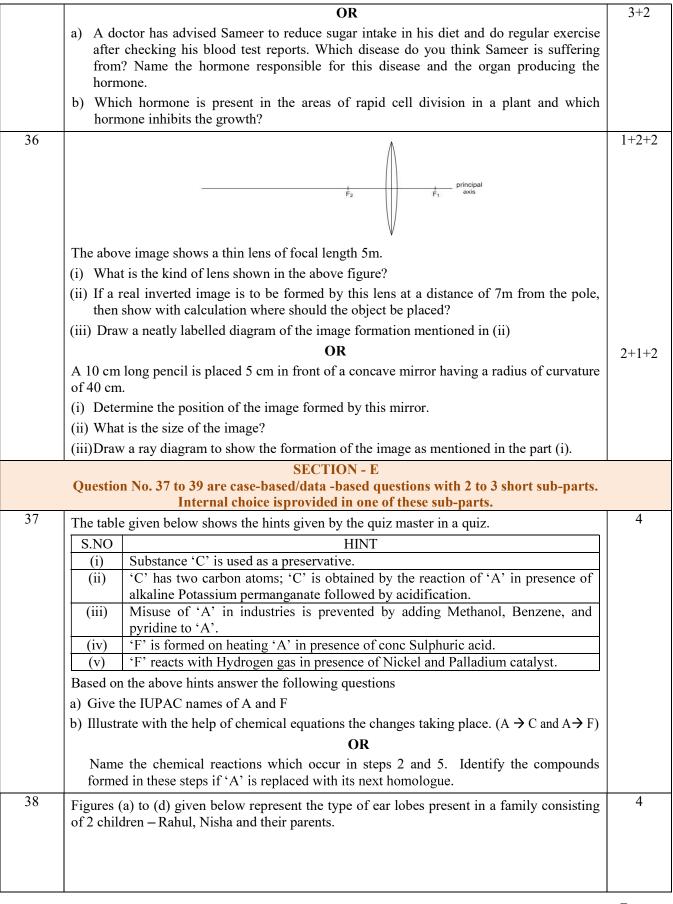
30		3
	What is the probability of a girl or a boy being born in a family? Justify your answer.	
31	(i) Explain why the refractive index of any material with respect to air is always greater 1.	1+1+1
	(ii) In the figure below a light ray travels from air into the semi-circular plastic block. Give a reason why the ray does not deviate at the semi-circular boundary of the plastic	
	block.	
	plastic block normal	
	40-1	
	light ray	
	(iii) Complete the ray diagram of the above scenario when the light ray comes out of the plastic block from the top flat end.	
32	(i) State the law that explains the heating effect of current with respect to the measurable properties in an electrical circuit.	2+1
	(ii) List the factors on which the resistance of a conductor depends.	
33	Anannya responded to the question: Why do electrical appliances with metallic bodies are connected to the mains through a three pin plug, whereas an electric bulb can be connected with a two pin plug?	2+1
	She wrote: Three pin connections reduce heating of connecting wires.	
	(i) Is her answer correct or incorrect? Justify.	
	(ii) What is the function of a fuse in a domestic circuit?	
	Section-D	
34	Question No. 34 to 36 are long answer questions.	5
	a) Rehmat classified the reaction between Methane and Chlorine in presence of sunlight as a substitution reaction. Support Rehmat's view with suitable justification and illustrate the reaction with the help of a balanced chemical equation.	Ü
	b) Chlorine gas was prepared using electrolysis of brine solution. Write the chemical equation to represent the change. Identify the other products formed in the process and give one application of each.	
	OR	
	Raina while doing certain reactions observed that heating of substance 'X' with vinegar like smell with a substance 'Y' (which is used as an industrial solvent) in presence of conc.	
	Sulphuric acid on a water bath gives a sweet-smelling liquid 'Z' having molecular formula	
	C <sub>4</sub> H <sub>8</sub> O <sub>2</sub> . When heated with caustic soda (NaOH), 'Z' gives back the sodium salt of and the	
	· · · · · · · · · · · · · · · · · · ·	
	compound 'Y'.	
	· · · · · · · · · · · · · · · · · · ·	
35	compound 'Y'.  Identify 'X', 'Y', and 'Z'. Illustrate the changes with the help of suitable chemical	5
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7











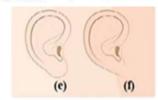




a) Rahul's Father

b) Rahul

c) Rahul's Mother d) Rahul's sister Nisha



Type of ear lobes

Excited by his observation of different types of ear lobes present in his family, Rahul conducted a survey of the type of ear lobes found {Figure (e) and (f)} in his classmates. He found two types of ear lobes in his classmates as per the frequency given below:

Sex	Free	Attached
Male	36	14
Female	31	19

On the basis of above data answer the following questions.

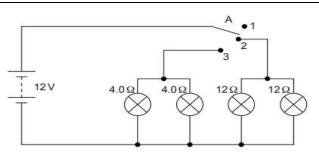
- a) Which of the two characteristics 'free ear lobe' or 'attached ear lobe' appears to be dominant in this case? Why?
- b) Is the inheritance of the free ear lobe linked with sex of the individual? Give reason for your answer.
- c) What type of ear lobe is present in father, mother, Rahul and his sister Nisha? Write the genetic constitution of each of these family members which explains the inheritance of this character in this family?

(Gene for Free ear lobe is represented by F and gene for attached ear lobe is represented by f for writing the genetic constitution).

#### OR

Suresh's parents have attached earl obes. What type of ear lobe can be seen in Suresh and his sister Siya? Explain by giving the genetic composition of all.

39



Vinita and Ahmed demonstrated a circuit that operates the two headlights and the two sidelights of a car, in their school exhibition. Based on their demonstrated circuit, answer the following questions.

- (i) State what happens when switch A is connected to
  - a) Position 2
  - b) Position 3
- (ii) Find the potential difference across each lamp when lit.
- (iii) Calculate the current

8

1+1+2













- a) in each 12  $\Omega$  lamp when lit.
- 6) In each 4  $\Omega$  lamp when lit.

OR

(iv) Show, with calculations, which type of lamp, 4.0  $\Omega$  or 12  $\Omega$ , has the higher power.













# Courses Offered

M.P.C.

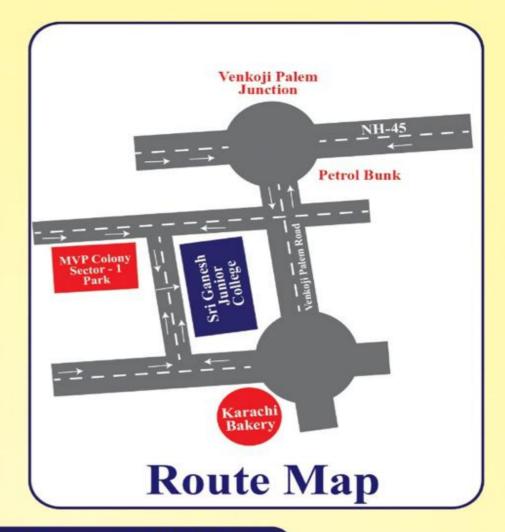
M- JEE, EAPCET

(Bİ.P.C.)

**NEET** - EAPCET

Long Term

MEET - IIT - EAPCET



**Test Streams & Subjects** 

M.P.C

MATHS 20Q PHYSICS 20Q CHEMISTRY 20O TOTAL 240 Marks BI.P.C
BIOLOGY 20Q
PHYSICS 20Q
CHEMISTRY 20O

For Each Wrong Answer -



For Each correct Answer +

+4

## SYLLABUS FOR SCHOLARSHIP TEST

MATHS

Polynomials, Quadratic equation, Pair of Linear equation, Arithmetic progressions, Coordinate geometry, Introduction to Trigonometry, Applications of Trignometry, Probability

PHYSICS

Motion, Force, Gravitation, Sound, Reflection of light, Work power and Energy, Refraction of light, Human Eye

CHEMISTRY

Some basic concepts of chemistry, Atomic structure, Chemical bonding, Solutions, Basic organic chemistry, Classification of elements and periodicity in Properties.

BIOLOGY

Diversity in living organisms, Natural Resources, Why do we fall ill, Nutrition, Excretion, Respiration, Transportation in Animals, Transportation in Plants.



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